

Transfer matrix method

1 One dimensional lattice gas

Real interacting gas in one-dimension can be approximated by partitioning a line of length L into M cells of length $a = L/M$. We will moreover assume that the line has periodic boundary conditions and the cells are sufficiently small to only include 0 or 1 particles. We neglect the kinetic energy of the particles and assume that two particles in two nearby cells have an energy of $-\varepsilon$. The gas is found at a temperature T and chemical potential μ . The goal is compute *exactly* the grand-potential $\Omega(L, T, \mu)$, the entropy, and the pressure of the gas.

Solution We designate the index of the cells by α and the occupation of each cell by $n_\alpha \in \{0, 1\}$. Then the grand partition function is given by

$$\Xi = \sum_{N=0}^M e^{\beta\mu N} \sum'_{\{n_\alpha\}} e^{\beta\varepsilon \sum_\alpha n_\alpha n_{\alpha+1}}, \quad (1)$$

where \sum' designates the constraint, $\sum_\alpha n_\alpha = N$. We can write it as an unconstrained sum,

$$\Xi = \sum_{\{n_\alpha\}} \exp \left[\beta\mu \sum_\alpha n_\alpha + \beta\varepsilon \sum_{\alpha=1}^M n_\alpha n_{\alpha+1} \right]. \quad (2)$$

Unlike the partition functions we saw previously, this is an interacting partition function, which can be solved exactly in a handful of cases. Luckily this is one of them. We note that we can write

$$\begin{aligned} \exp \left[\beta\mu \sum_\alpha n_\alpha + \beta\varepsilon \sum_{\alpha=1}^M n_\alpha n_{\alpha+1} \right] &= \exp \left[\beta\mu \frac{(n_1 + n_2)}{2} + \beta\varepsilon n_1 n_2 \right] \exp \left[\beta\mu \frac{(n_2 + n_3)}{2} + \beta\varepsilon n_2 n_3 \right] + \dots \\ &\quad (3) \\ &\quad \times \exp \left[\beta\mu \frac{(n_M + n_1)}{2} + \beta\varepsilon n_M n_1 \right], \end{aligned}$$

where the last term uses the fact that the line is closed into a ring. We further note that for example for $M = 3$, the product

$$\sum_{n_1 n_2 n_3} \exp \left[\beta\mu \frac{(n_1 + n_2)}{2} + \beta\varepsilon n_1 n_2 \right] \exp \left[\beta\mu \frac{(n_2 + n_3)}{2} + \beta\varepsilon n_2 n_3 \right] \exp \left[\beta\mu \frac{(n_3 + n_1)}{2} + \beta\varepsilon n_3 n_1 \right], \quad (4)$$

can be written as the trace of three identical matrices,

$$\text{Tr } T^3, \quad (5)$$

where

$$T_{n_1 n_2} = \exp \left[\beta \mu \frac{(n_1 + n_2)}{2} + \beta \varepsilon n_1 n_2 \right] = \begin{pmatrix} 1 & e^{\frac{1}{2}\beta\mu} \\ e^{\frac{1}{2}\beta\mu} & e^{\beta(\mu+\varepsilon)} \end{pmatrix}. \quad (6)$$

Therefore for generic M we have,

$$\Xi = \text{Tr } T^M = \text{Tr } U U^{-1} T^M = \text{Tr } U^{-1} T^M U = \lambda_1^M + \lambda_2^M \quad (7)$$

where in the last two equalities we used U the matrix which diagonalizes T , and the cyclic property of the trace. The fundamental equation for T is given by,

$$\lambda^2 - \left(1 + e^{\beta(\mu+\varepsilon)}\right) \lambda + \left(e^{\beta(\mu+\varepsilon)} - e^{\beta\mu}\right) = 0 \quad (8)$$

Therefore,

$$\begin{aligned} \lambda_{1,2} &= \frac{1 + e^{\beta(\mu+\varepsilon)}}{2} \pm \sqrt{\frac{(1 - e^{\beta(\mu+\varepsilon)})^2}{4} + e^{\beta\mu}} \\ &= e^{\frac{\beta(\mu+\varepsilon)}{2}} \left\{ \cosh \frac{\beta}{2} (\mu + \varepsilon) \pm \sqrt{\sinh^2 \frac{\beta}{2} (\mu + \varepsilon) + e^{-\beta\varepsilon}} \right\}, \end{aligned} \quad (9)$$

in the thermodynamic limit we have,

$$-P = \frac{\Omega}{L} = \lim_{M \rightarrow \infty} \frac{-\frac{1}{\beta} \ln \Xi}{Ma} = -\frac{(\mu + \varepsilon)}{2} - \frac{1}{\beta} \ln \left\{ \cosh \frac{\beta}{2} (\mu + \varepsilon) + \sqrt{\sinh^2 \frac{\beta}{2} (\mu + \varepsilon) + e^{-\beta\varepsilon}} \right\}, \quad (10)$$

where we used $\Omega = -PV$, which follows from the fundamental equation of thermodynamics. Therefore the pressure of the gas is,

$$P(\mu, T) = \frac{(\mu + \varepsilon)}{2} + \frac{1}{\beta} \ln \left\{ \cosh \frac{\beta}{2} (\mu + \varepsilon) + \sqrt{\sinh^2 \frac{\beta}{2} (\mu + \varepsilon) + e^{-\beta\varepsilon}} \right\}, \quad (11)$$

and the entropy is,

$$S = - \left(\frac{\partial \Omega}{\partial T} \right)_{\mu}. \quad (12)$$

Particle number fluctuations (which is proportional to isothermal compressibility) are given by,

$$\frac{\beta}{L} \left[\langle N^2 \rangle - \langle N \rangle^2 \right] = \left(\frac{\partial^2 P}{\partial \mu^2} \right)_{T, M} \quad (13)$$

We would like to note in passing that Ω/L is analytic for $T > 0$, and therefore there is not phase-transition in this model except at $T = 0$.